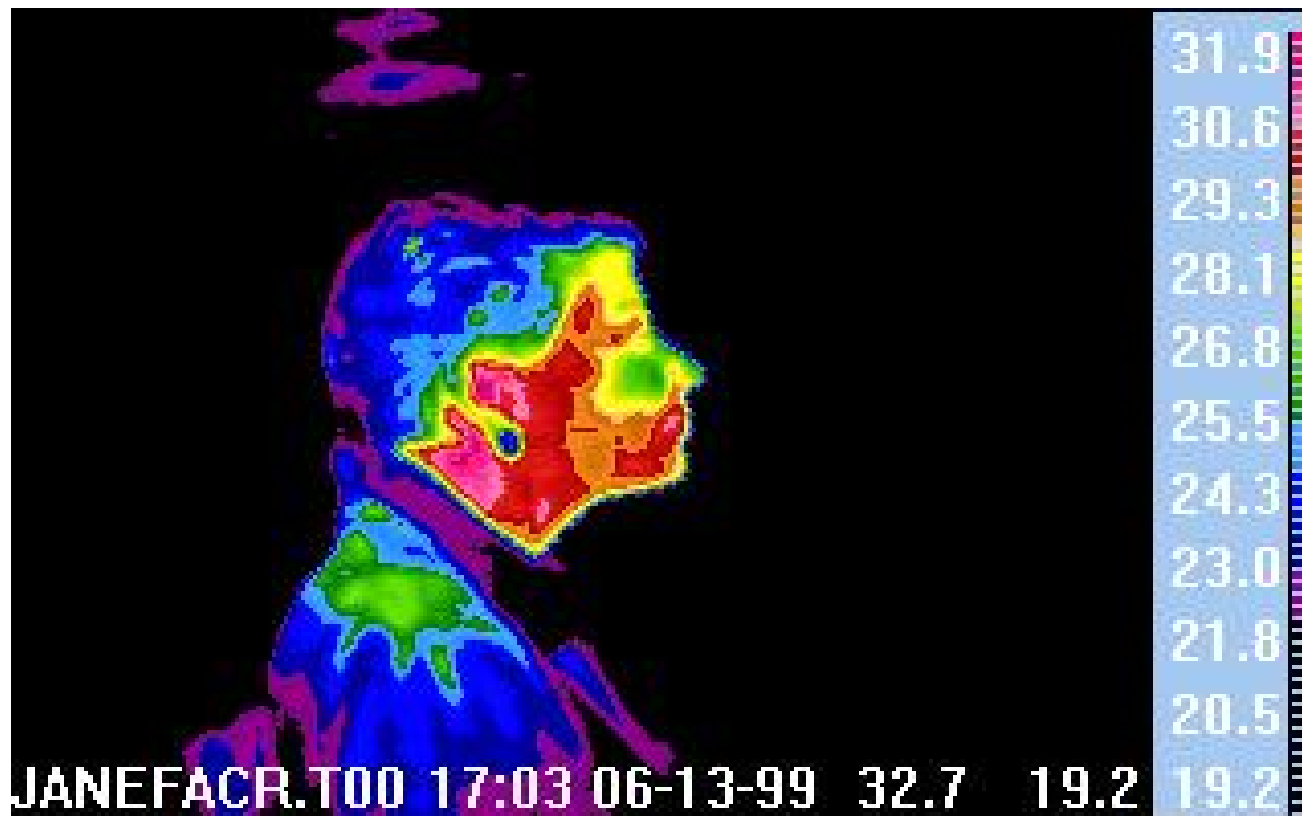


Thermal Infrared Systems

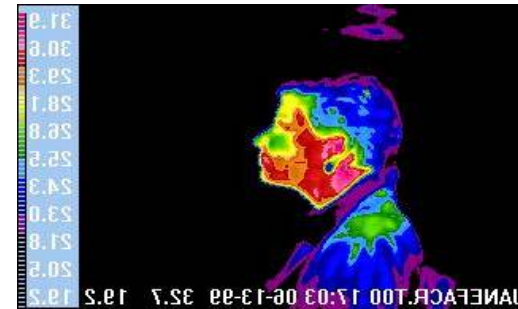
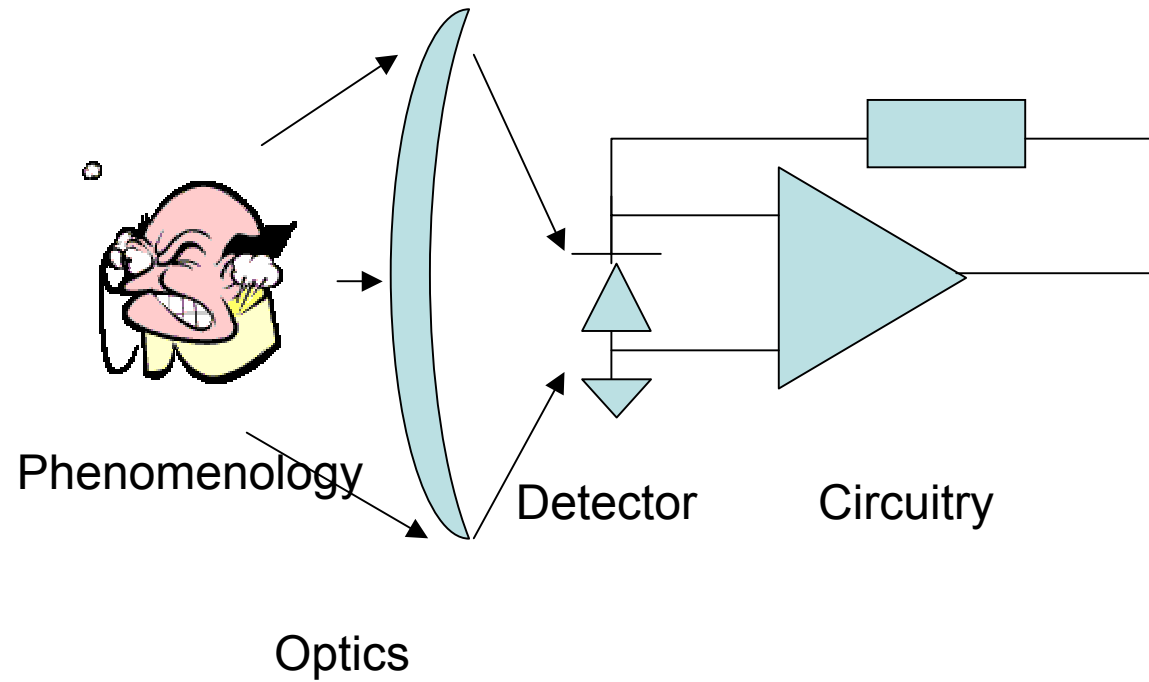
Lecturer: B T Yang 楊丙邨 March 2005 NTU



Lecture Outline

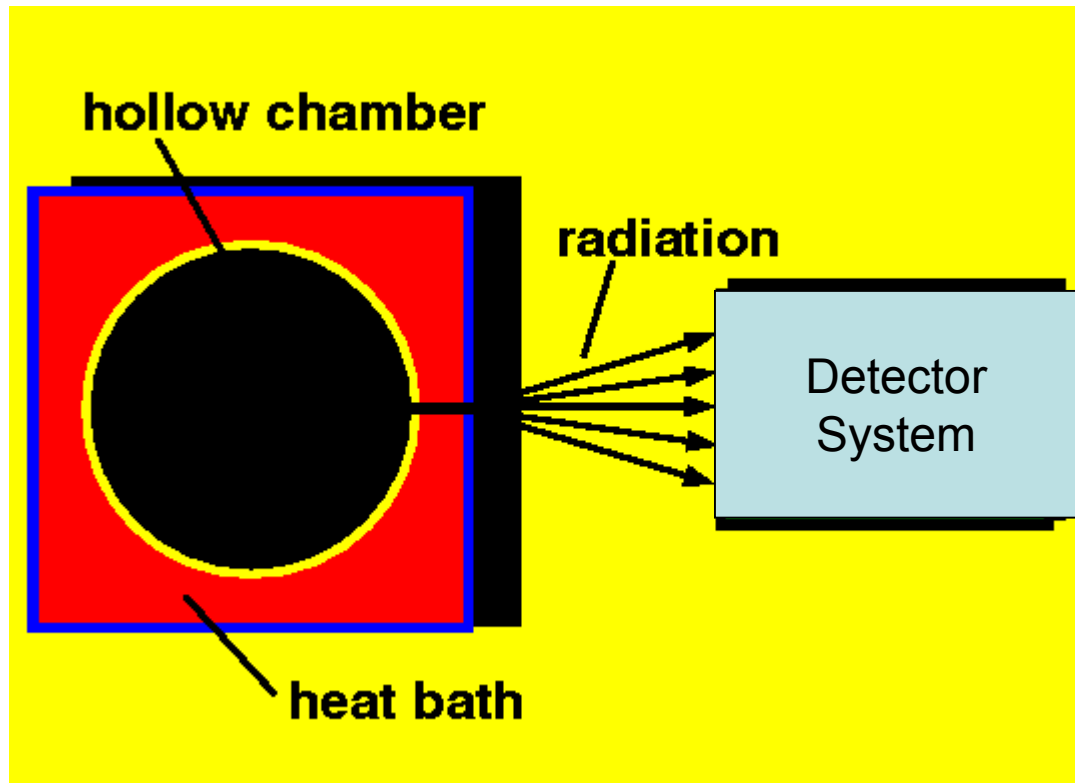
1. Phenomenology: “What is”
2. Optics
3. IR Detectors: Thermal, PC, PV
4. IR Detector Circuitry and Noises
5. IR Systems and Applications

Typical IR System



Signal Processing
and Display

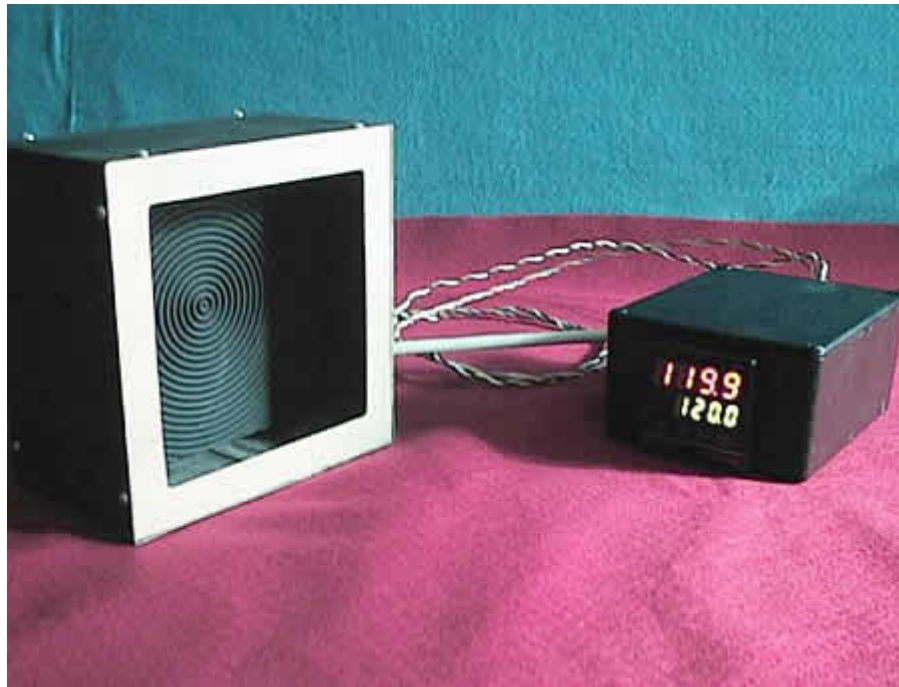
IR is Never Complete without Introducing the “Blackbody”



“Black” means No Light is Reflected, but “Light” can be emitted!

Grooved Planar Black Body Source

- “Grooved surface enhancing the emissivity

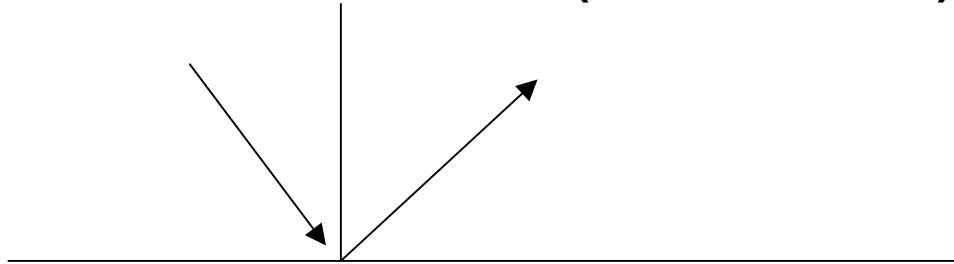


Definition of a Black Body

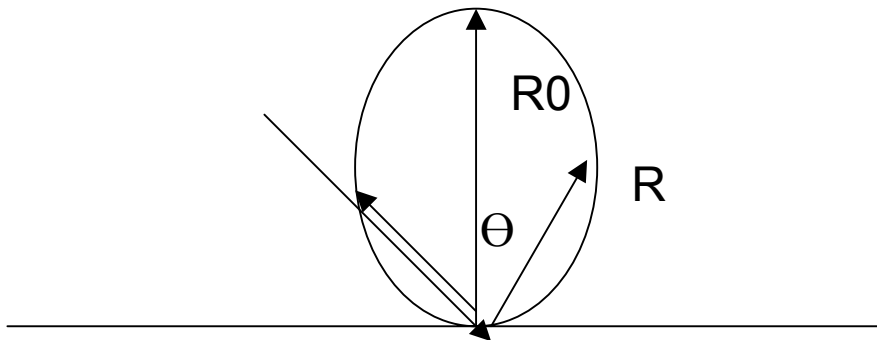
- **A blackbody absorbs all incident radiation; $r=0$**
- **At a given temperature, no surface can emit more energy than a blackbody**
- **A blackbody is a “diffuse” emitter that follows the “Lambertian Laws”**

Lambertian Law

- Specular Surface (reflective)



- Lambertian Surface (diffuse surface)

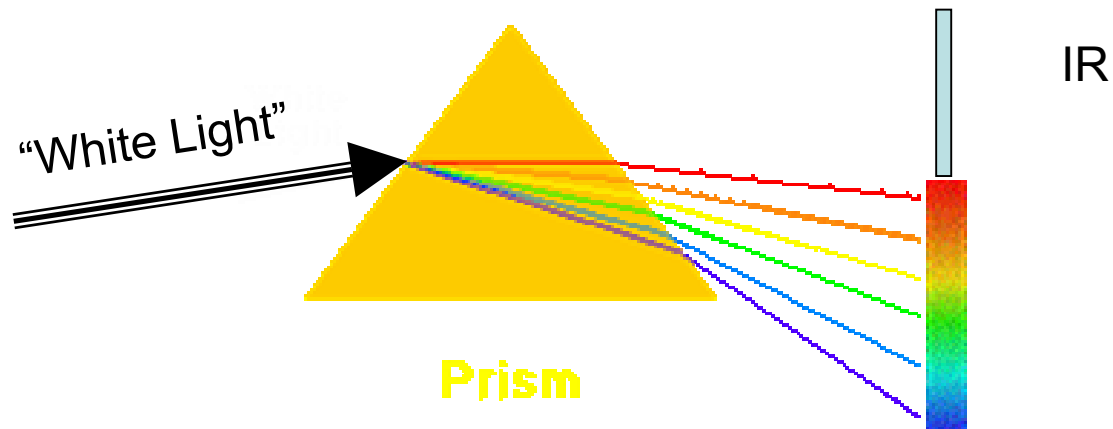


$$R = R_0 \cos \theta$$

The beginning of Infrared

Infra= Ln. below

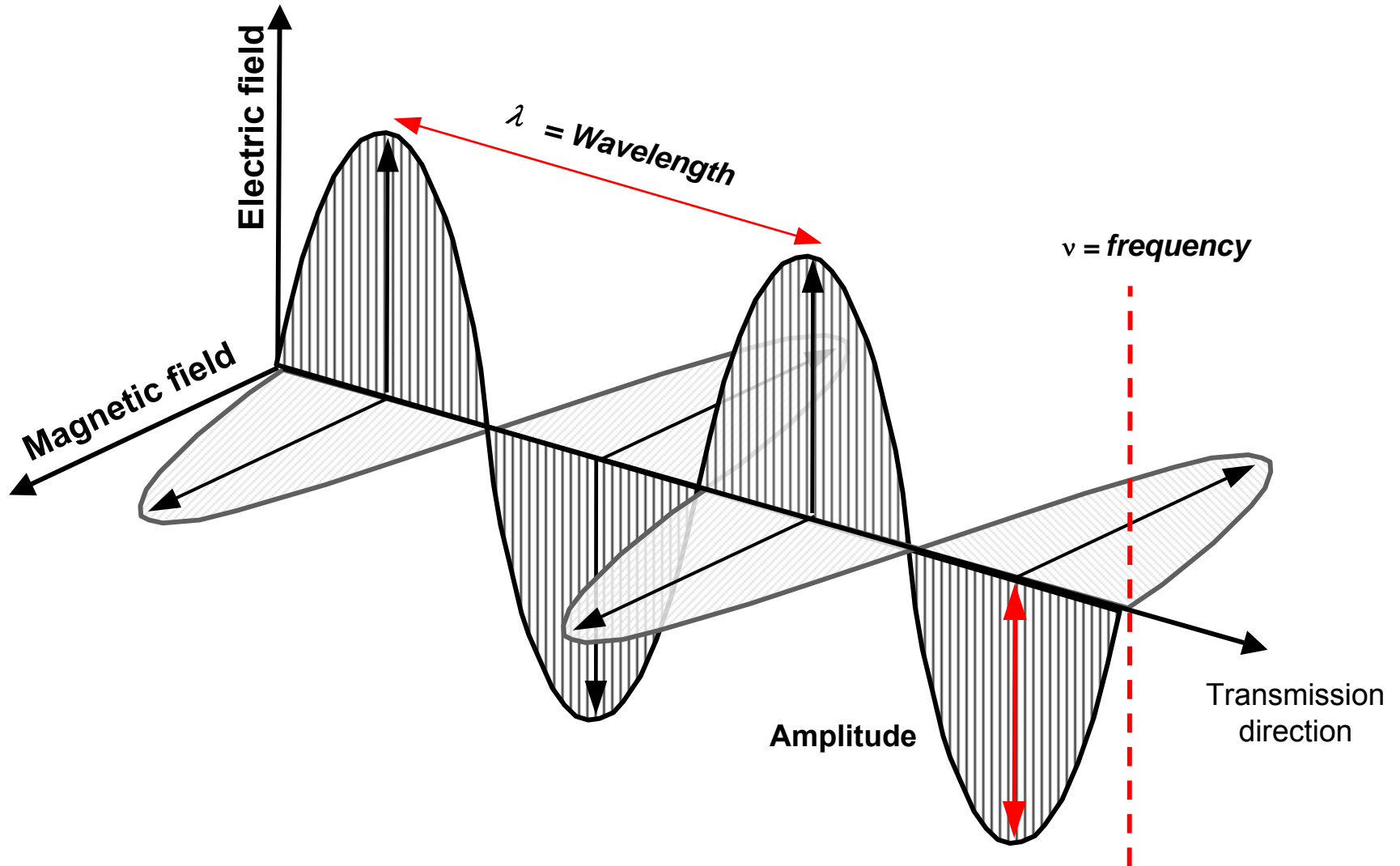
- In 1800, Sir William **Herschel**, using a prism to spread sunlight, observed the heating “beyond the red end” of the visible light spectrum



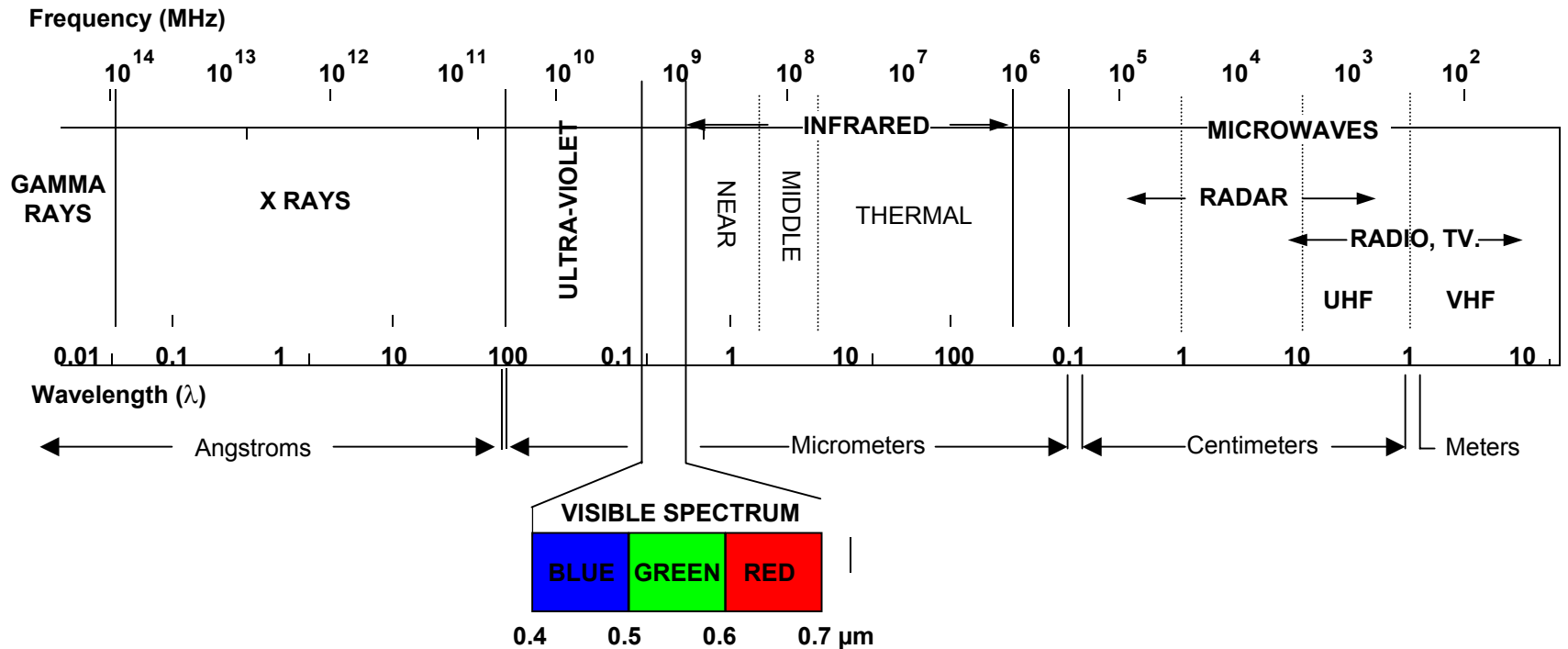
IR: Heat?

- A known effect of infrared light on skin is dilation of blood vessels that transport blood to and from the skin for cooling=> sensation of heat!
- According to Kirchhoff Law, if $r=0$
$$\varepsilon(\text{absorptivity}) = \sigma(\text{emissivity})$$
- Since skin is a good IR emitter then it must be a good IR absorber!

Light: An Electromagnetic wave



The Electromagnetic Spectrum



IR Frequency and Energy

- Frequencies: $.003 \times 10^{14}$ to 4×10^{14} Hz
- Wavelengths: 1 mm – $0.7 \mu\text{m}$
- Quantum energies: 0.0012 - 1.65 eV

Planck's Equation

M_λ : Spectral Exitance [$\text{W} \cdot \text{cm}^{-2} \cdot \mu\text{m}^{-1}$]

λ : wavelength [μm]

T: absolute temperature [K]

h = Planck's constant = 6.63×10^{-34} W sec²

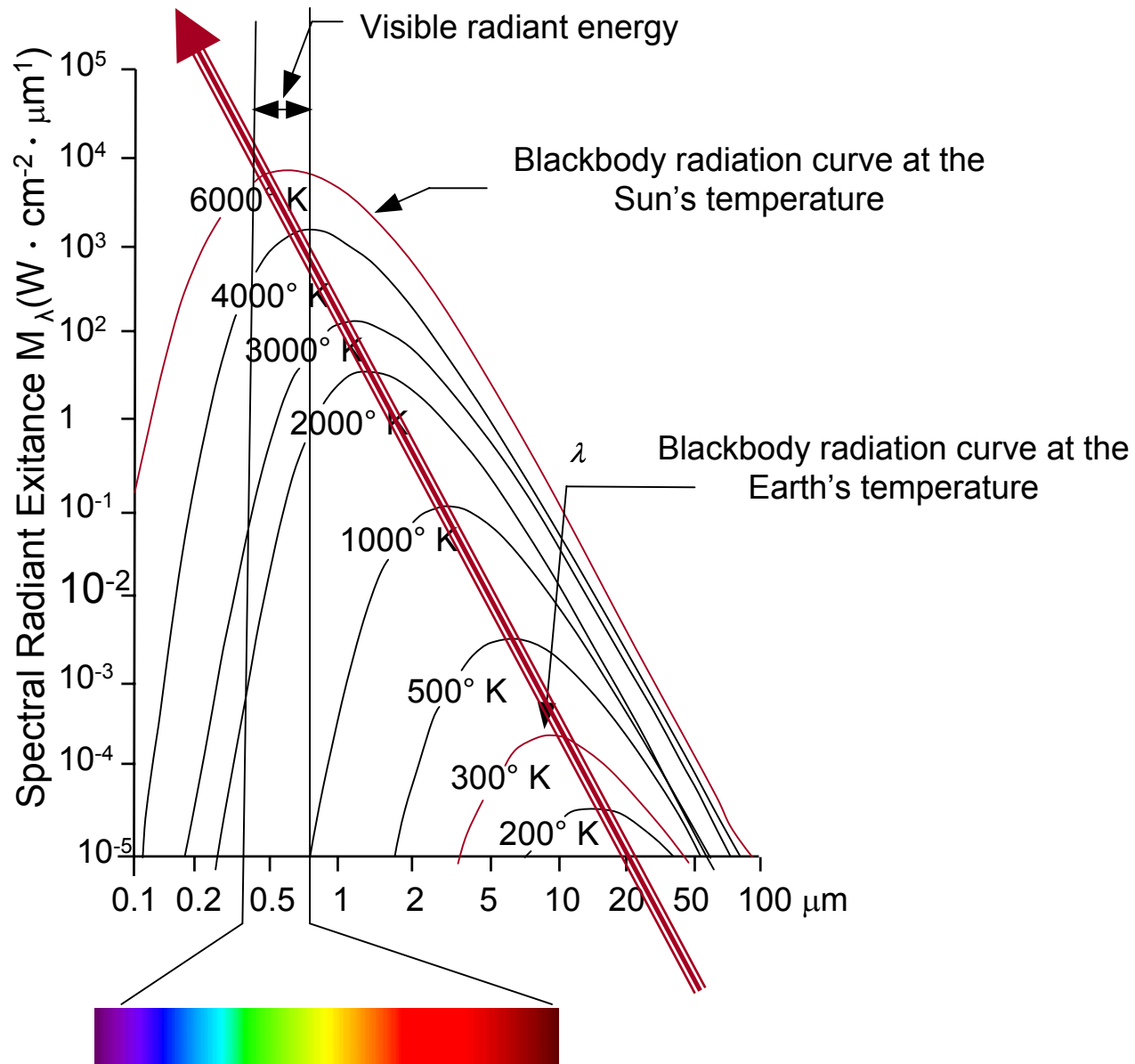
C = 3×10^{14} $\mu\text{m}/\text{sec}$

k = Boltzmann's constant 1.38×10^{-23} W sec K⁻¹)

$$M_\lambda (\lambda, T) = \frac{2 \pi h c^2}{\lambda^5} \frac{1}{e^{\frac{hc}{\lambda k T}} - 1}$$

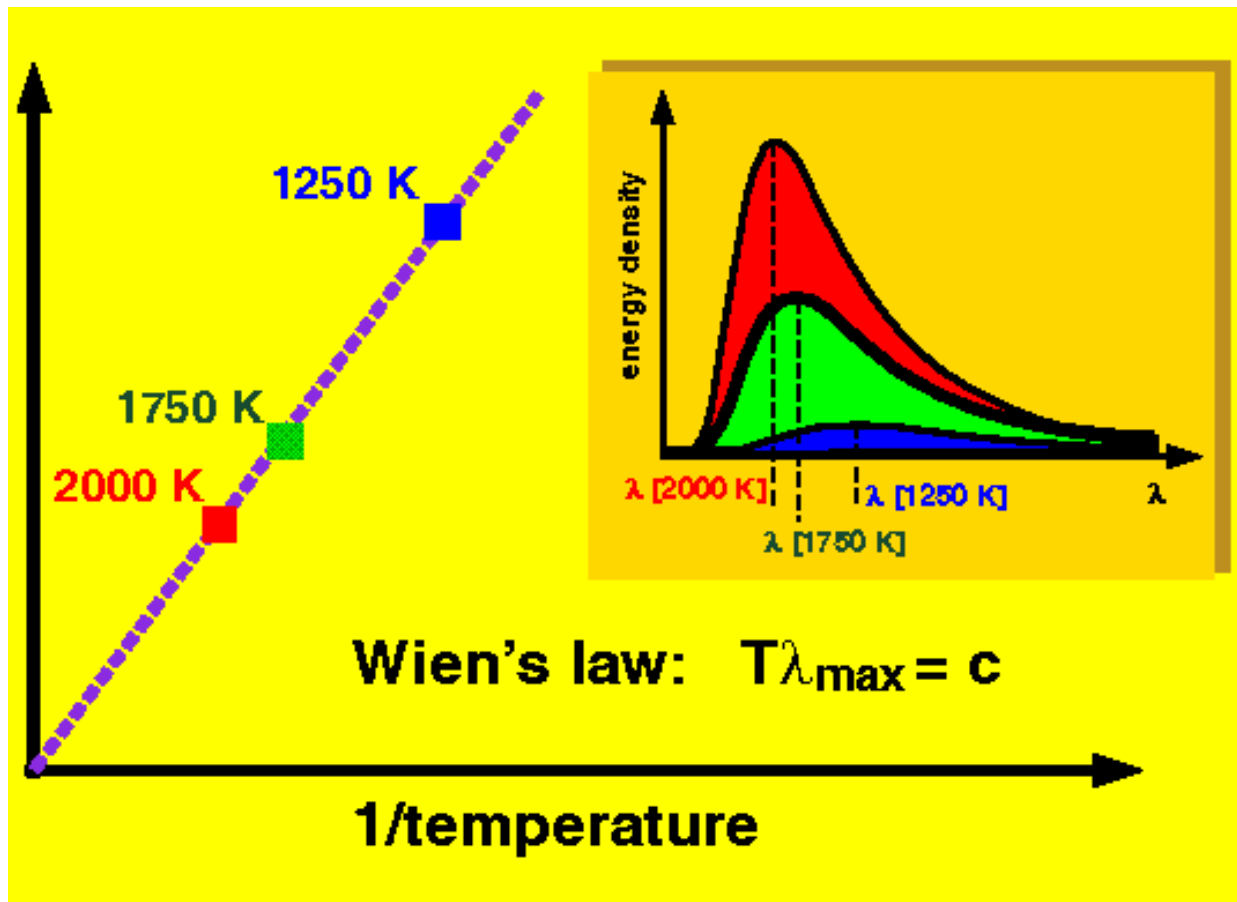
$$= \frac{3.74 \times 10^4}{\lambda^5 \left[e^{\frac{1.44 \times 10^4}{\lambda T}} - 1 \right]}$$

Spectral exitance of a blackbody



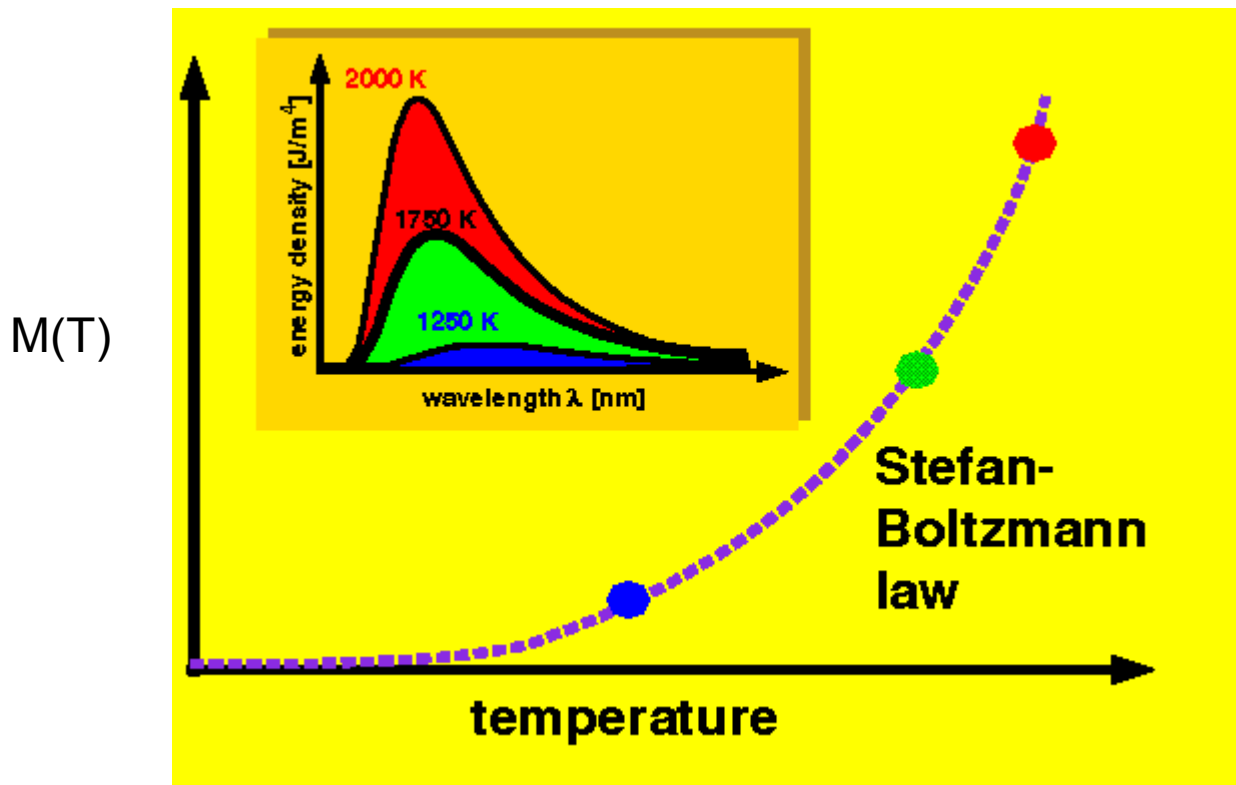
Wien's Law

- $\lambda_{\max} T \sim 3000 \mu\text{m}\cdot\text{K}$



Stefan-Boltzmann's Equation of Radiation

- $M(T) = \int M_\lambda(\lambda, T) d\lambda = \sigma T^4$ [W·cm⁻²]
- **M(T): Exitance (not Spectral Exitance)**
- σ : Stefan-Boltzmann's constant 5.67×10^{-12} W · cm⁻² · K⁻⁴



Grey Body?

- When emissivity ε is not unity
- Most physical surfaces are grey bodies
 $\varepsilon_{\text{skin}} \sim 0.95$, then it must be “Approximated as a Blackbody

$$M_{\lambda} = \varepsilon M_{\lambda}$$

$$M = \varepsilon \sigma T^4$$

Atmospheric Transmission Spectra

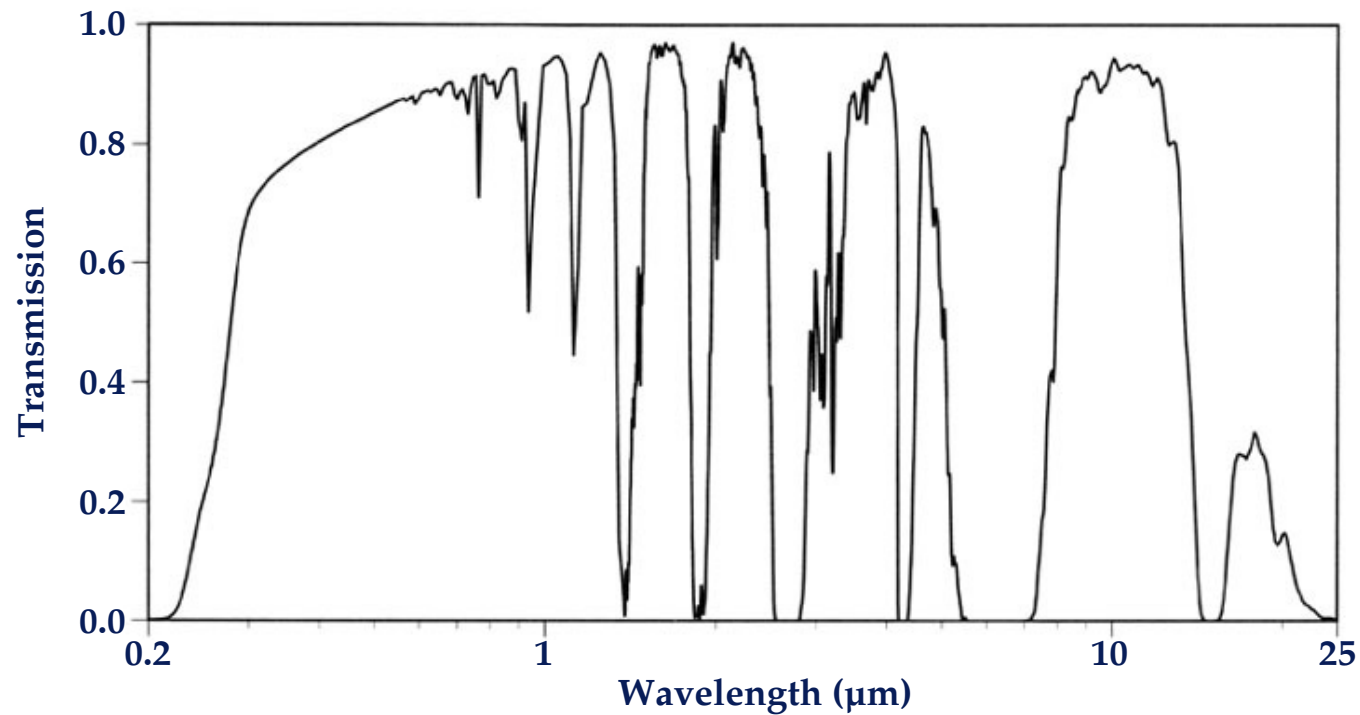
UV

VNIR

SWIR

MWIR

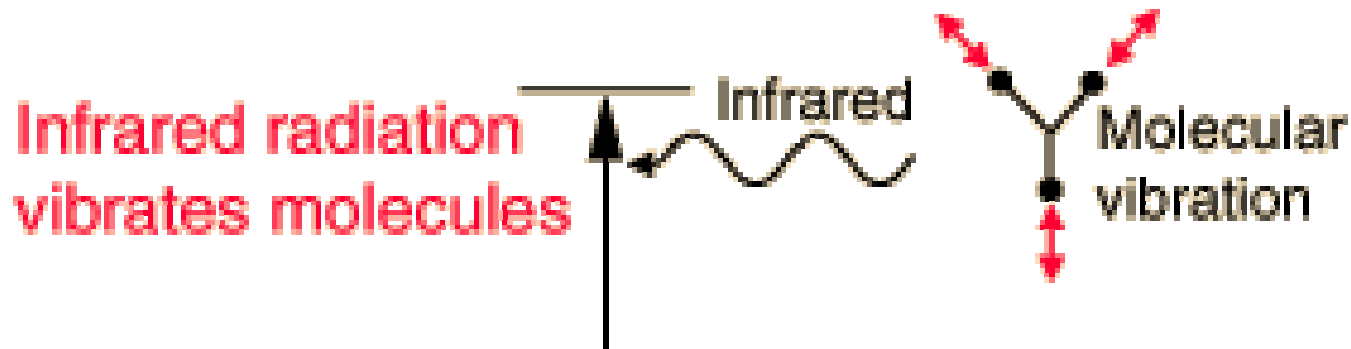
LWIR



Infrared Interactions

<http://hyperphysics.phy-astr.gsu.edu/hbase/mod3.html#c3>

- . The result of infrared absorption is heating of the tissue since it increases molecular vibrational activity..



•

Discrete Energy State

- Planck's 1900's "lucky Guess" $\Delta E = h\nu$

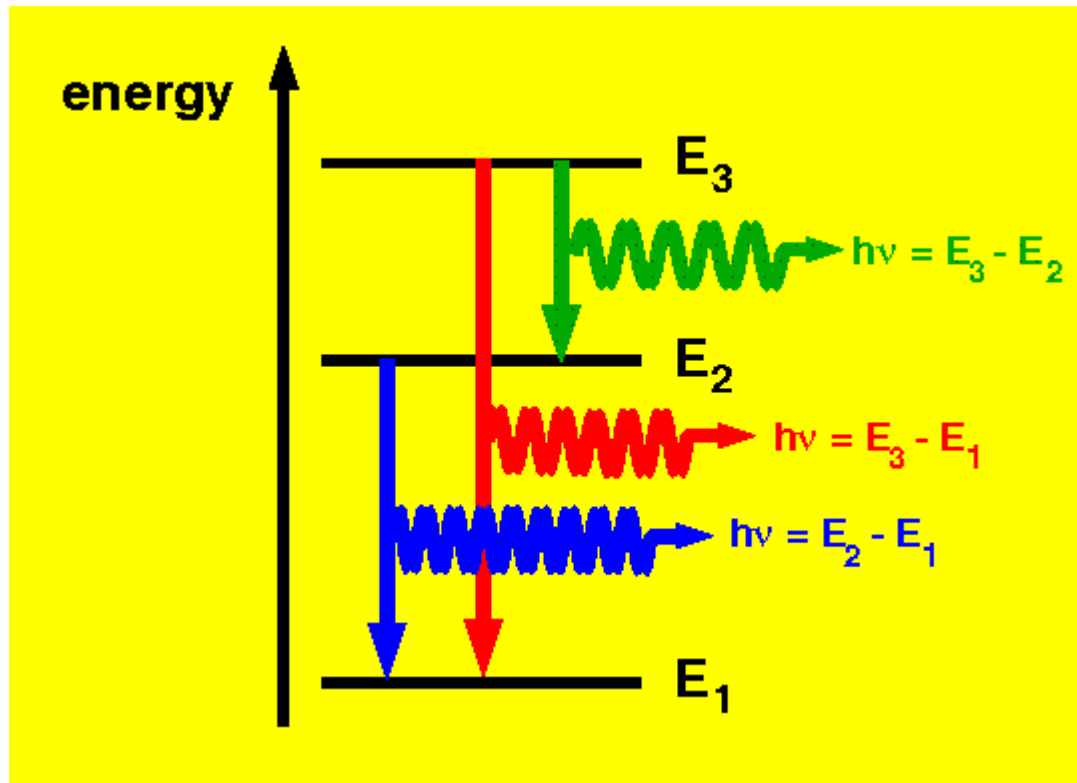
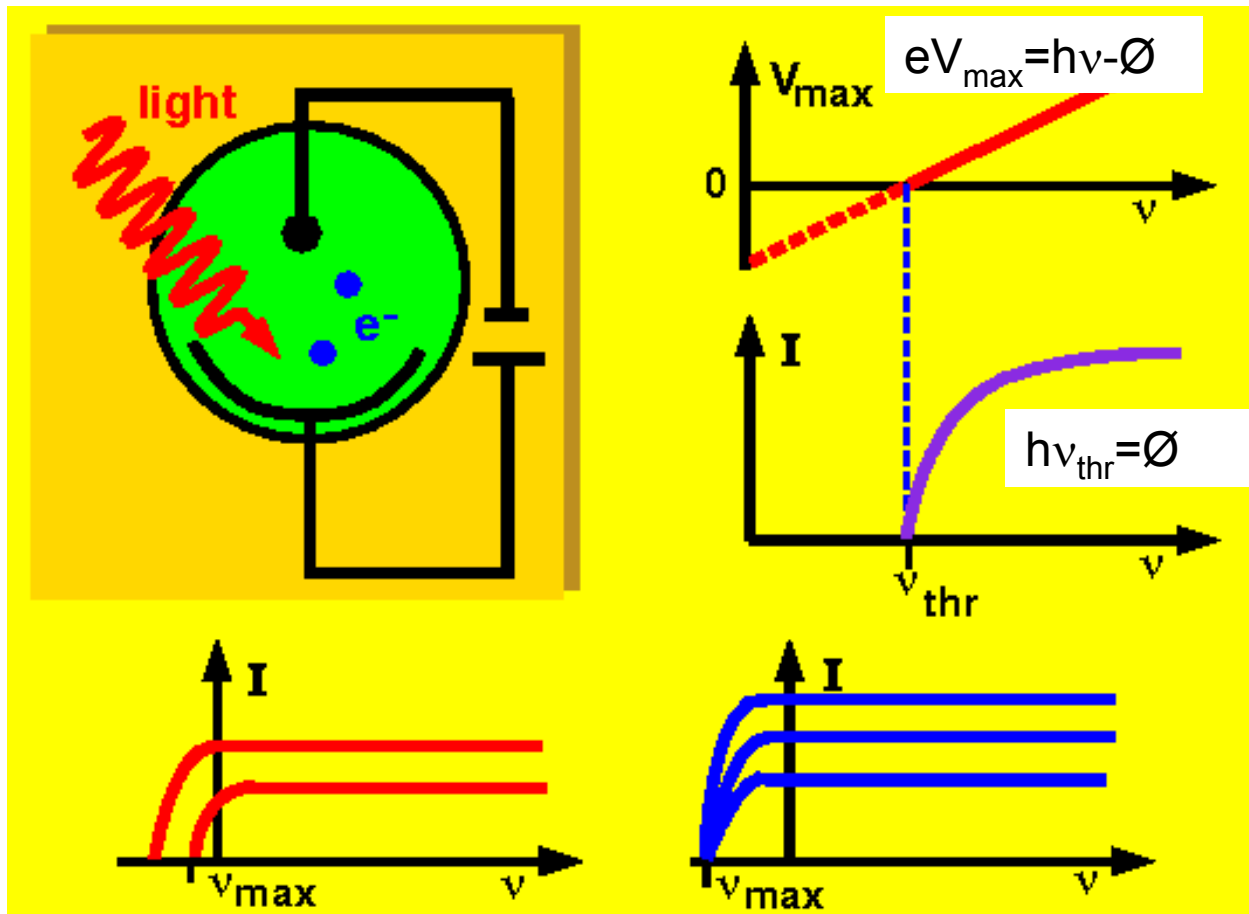


Photo-Electric Effect

- Einstein 1905's Paper Confirming the Discret Energy

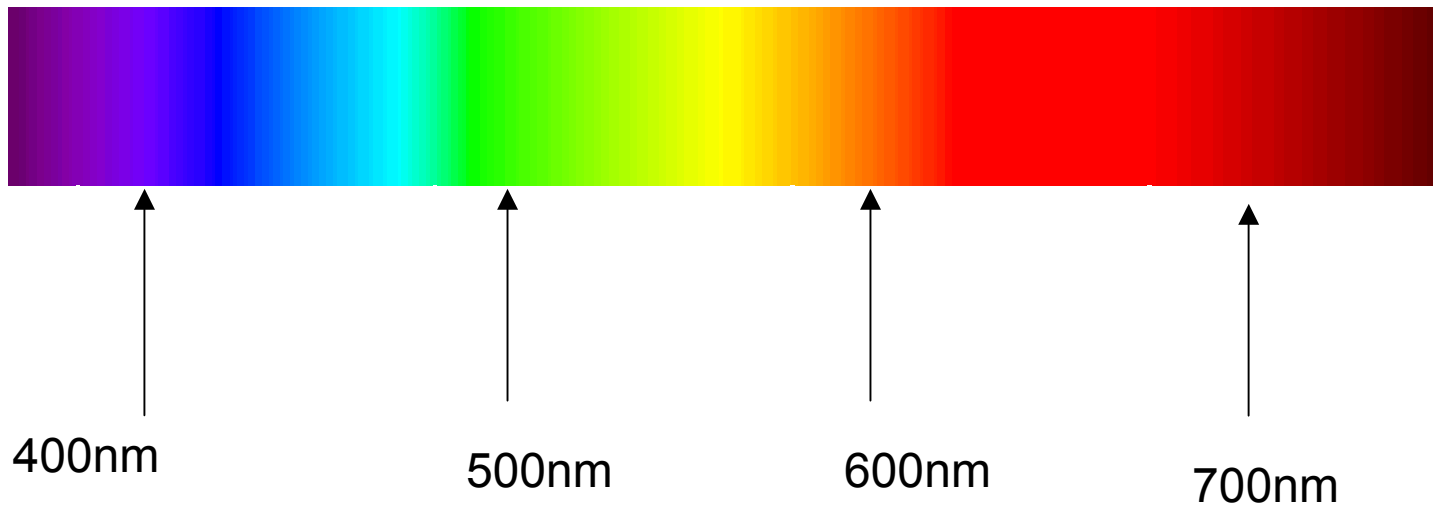


Varying ν

Same ν , varying intensity

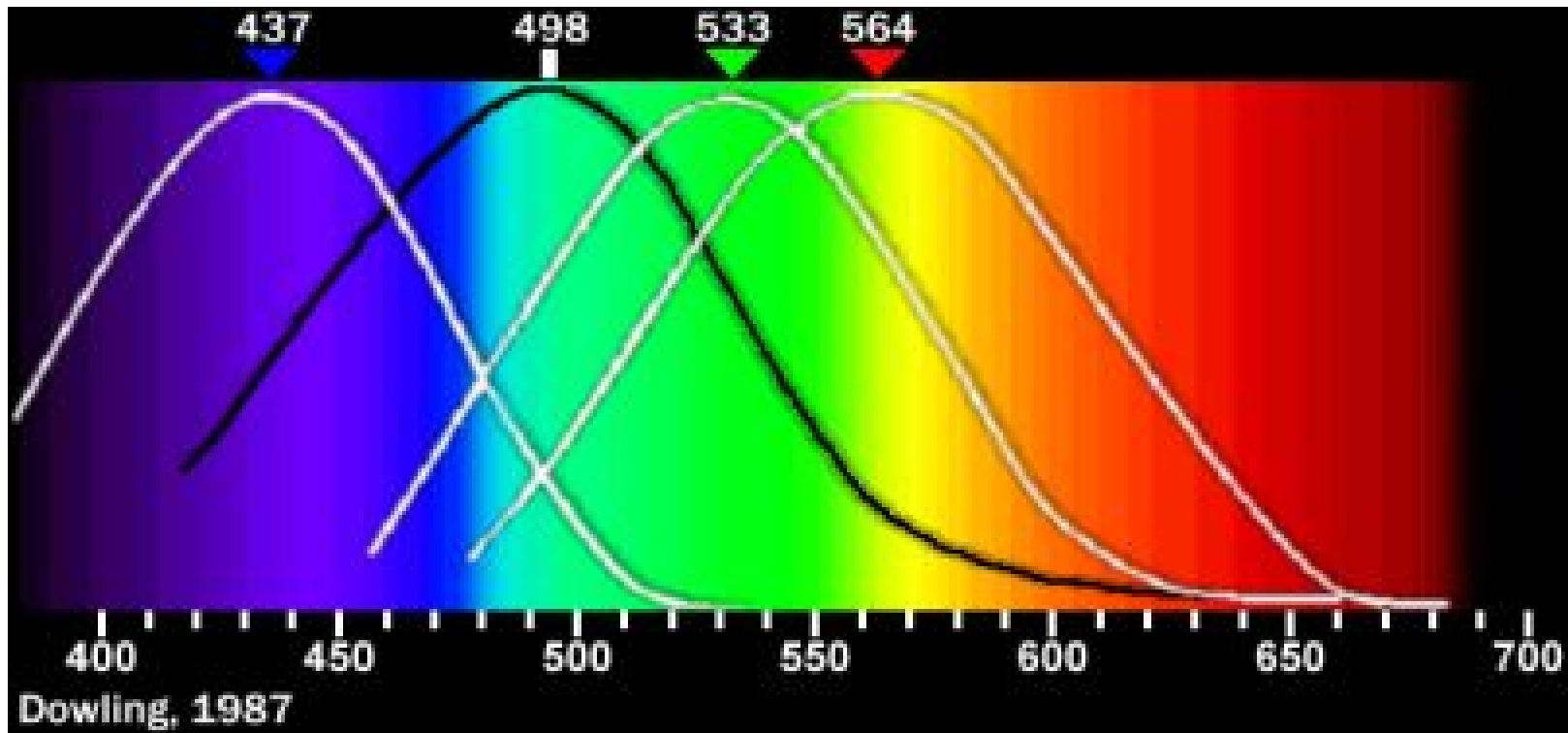
Visible Spectral Range

- Visible Band: 400nm to 700nm



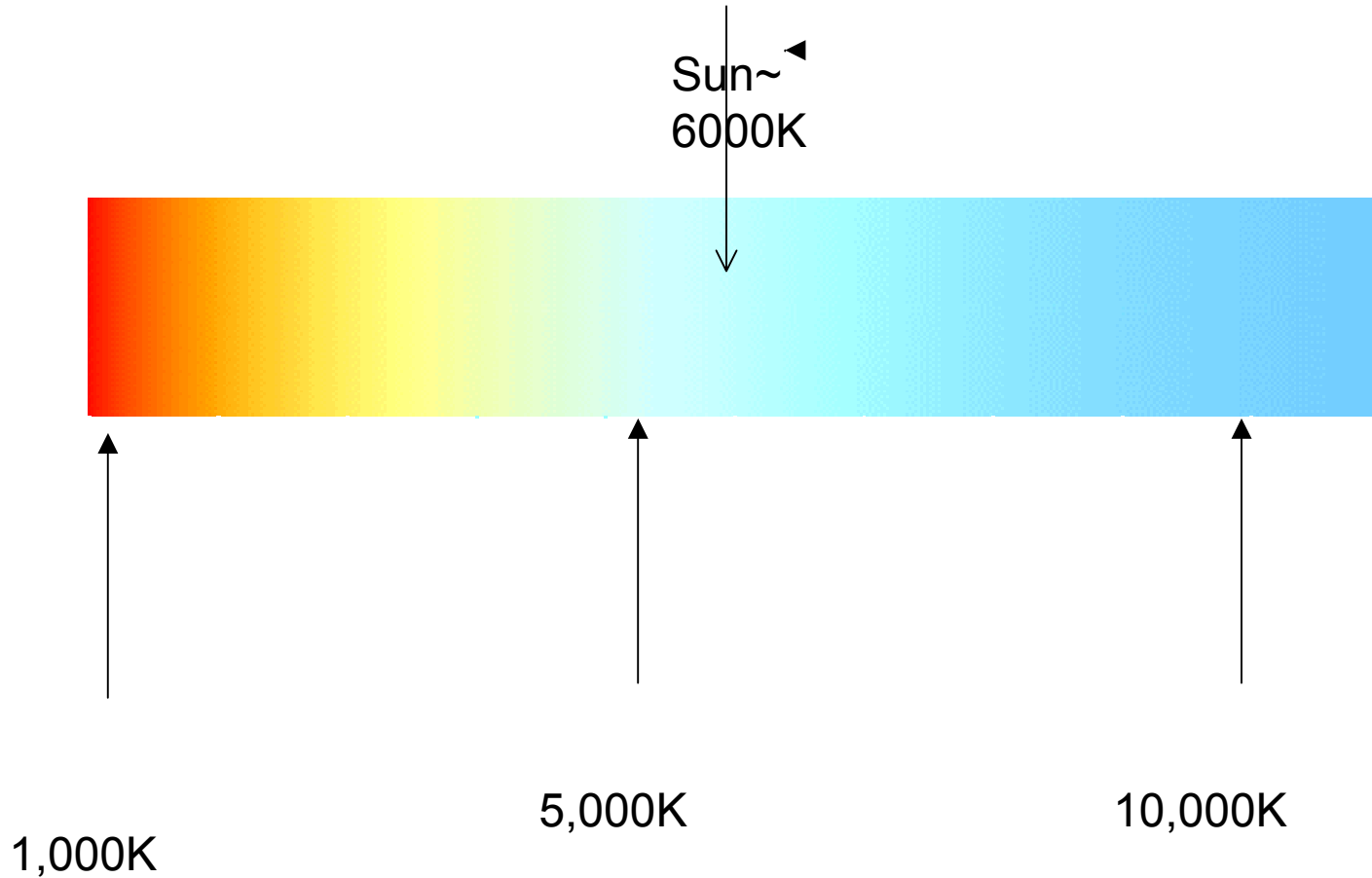
Eye's Cones' (3) and Rods' Responses

- Rods for night vision (more sensitive)
- Cones for color day vision



“Color” Temperature

An Apparent Visible Color of a Blackbody at T



Night Goggles are “not” true Thermal Images

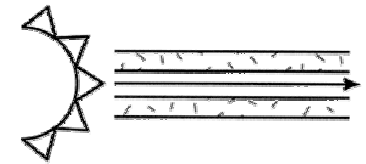
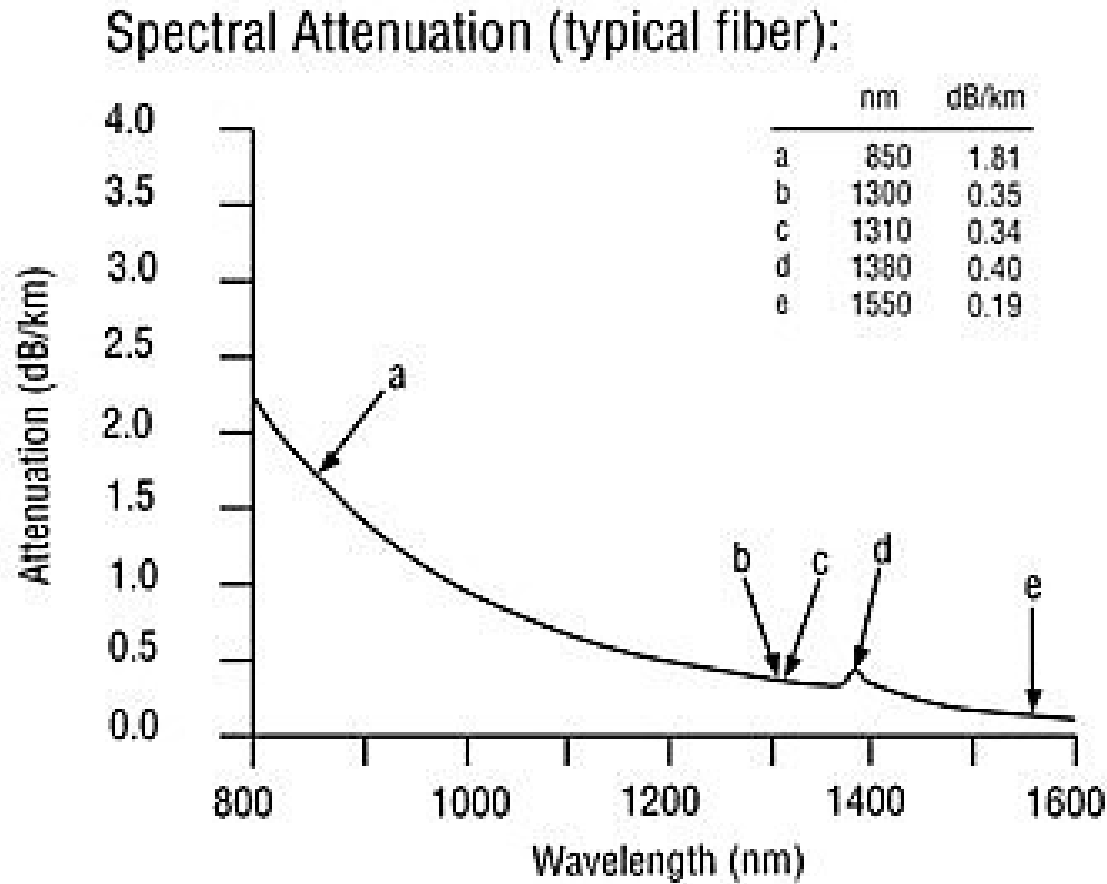
- Night Goggle Images are “Reflected NIR Images”, not “Emitted Thermal images”



Many Low-Cost Low-Light Detection Systems are NIR Systems

“Near IR Wavelength Used for Optical Communications”

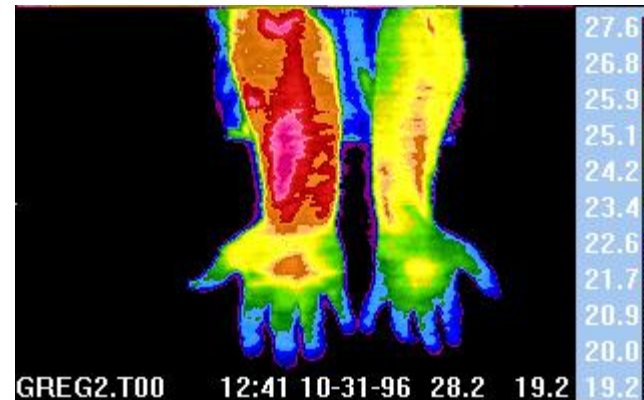
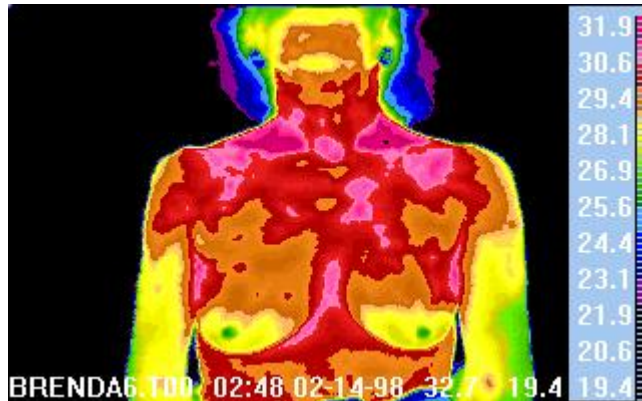
“Single mode fiber”
single path through the fiber



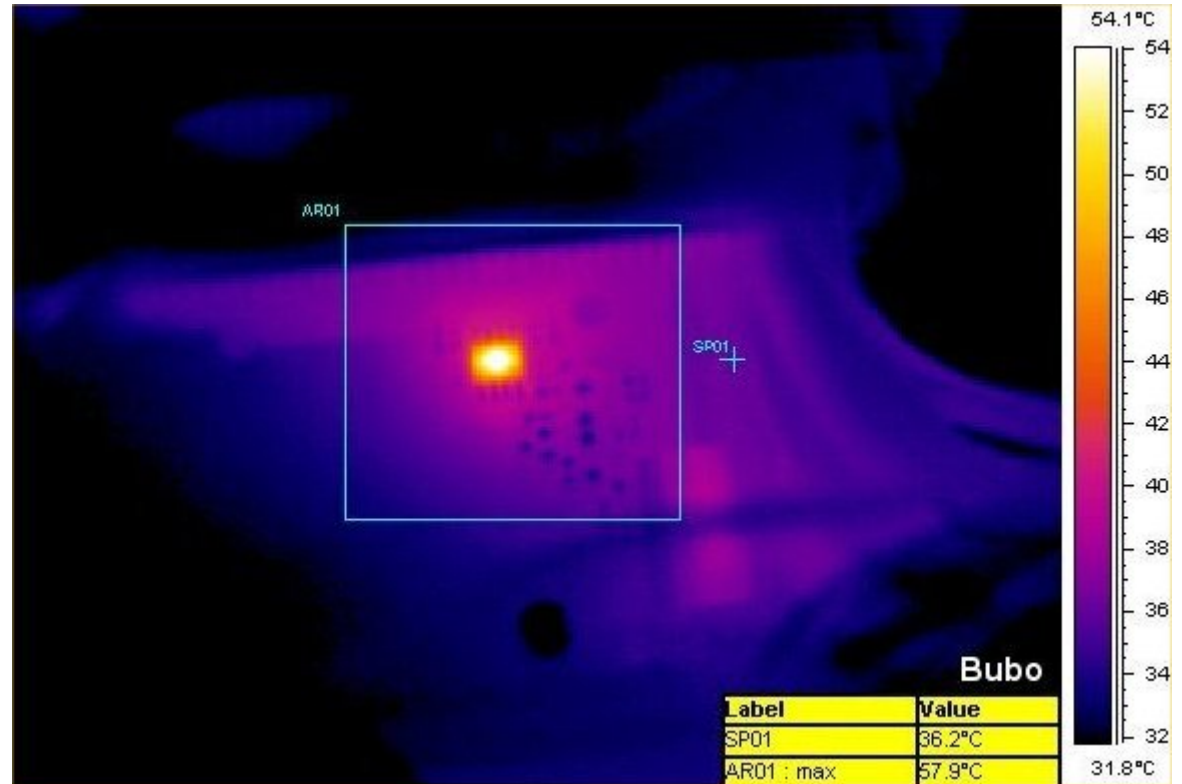
Singlemode Fibre 10/125

Human Thermal Images

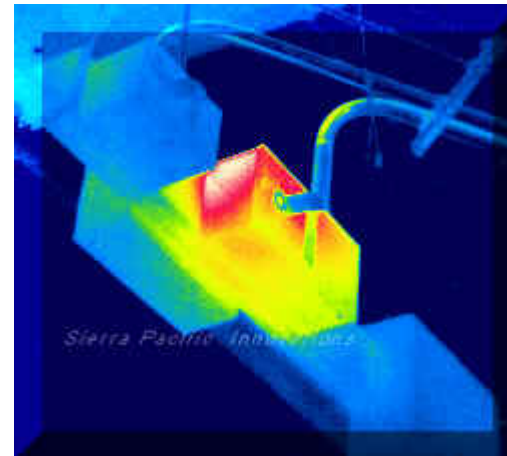
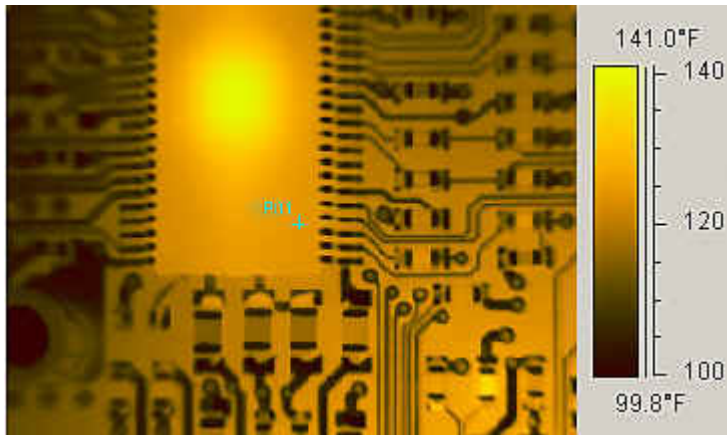
- http://www.ir55.com/infrared_IR_camera.html



PC Board Localized Heating

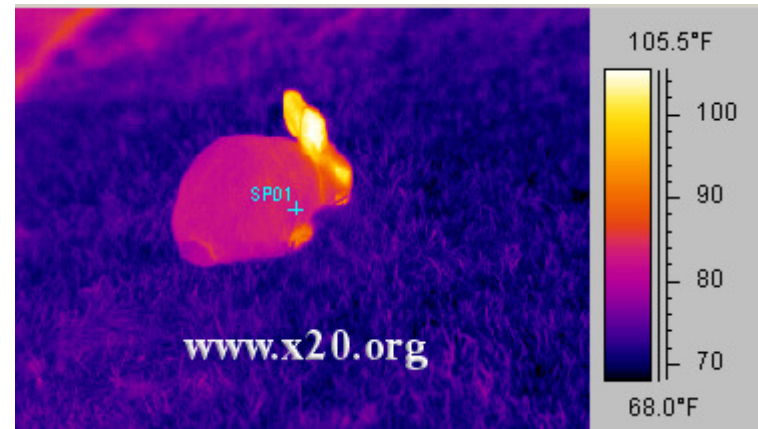
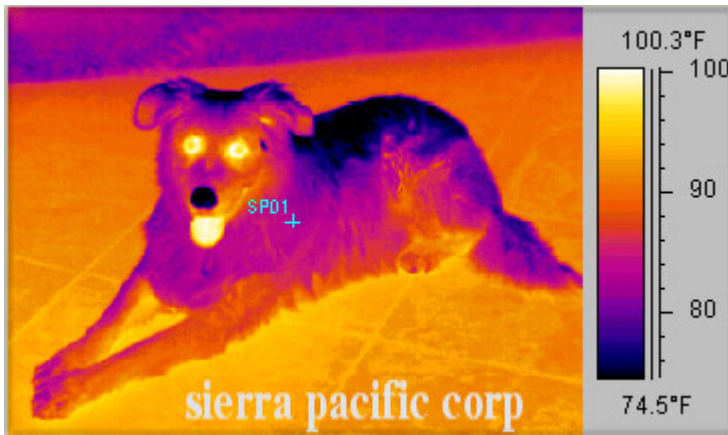


Localized IC Chip Detection



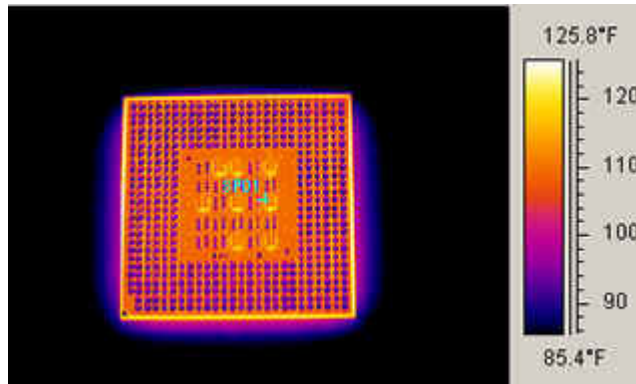
Burglar Detection

-



Underside Celeron Chip

-



SARS Temperature Screening

-



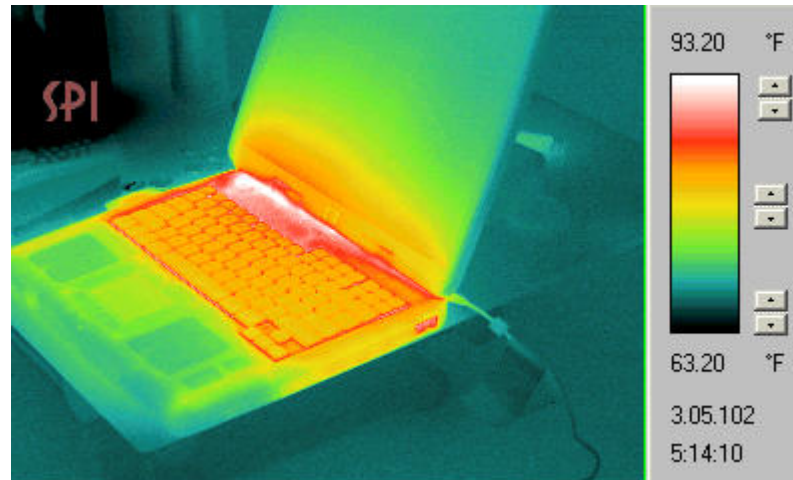
Preventive Maintenance

- Electrical Fuse Thermal Image

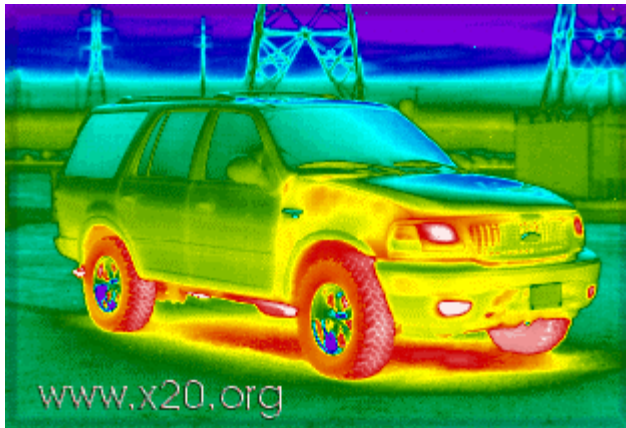


Thermal Management

-

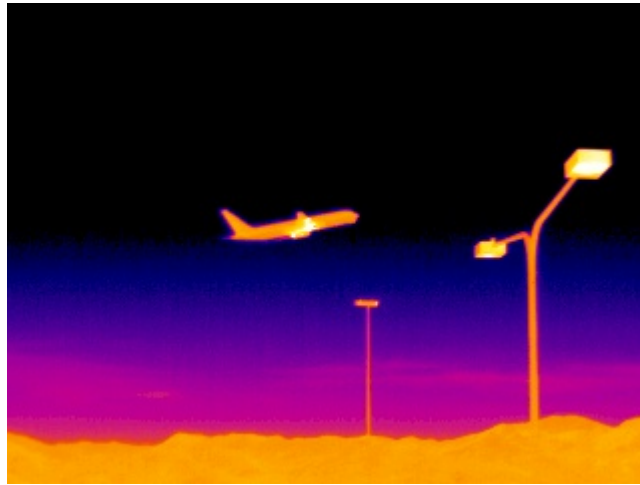


Defense Applications



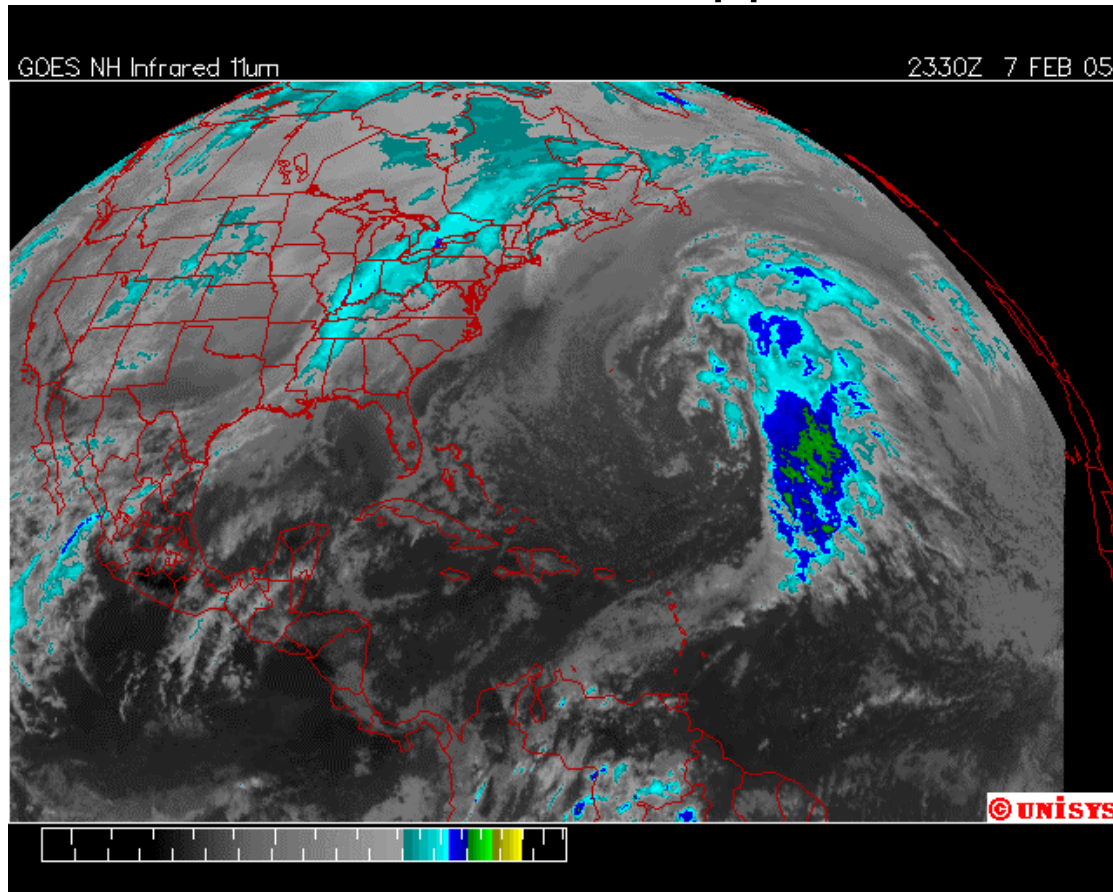
Sky Surveillance

- Collision Prevention



Weather Monitoring

- Geosynchronous Weather Satellite Application



What “Limits” Your Measurements?

1. Spatial (How Small an Area Can the System Resolved?):

Optics

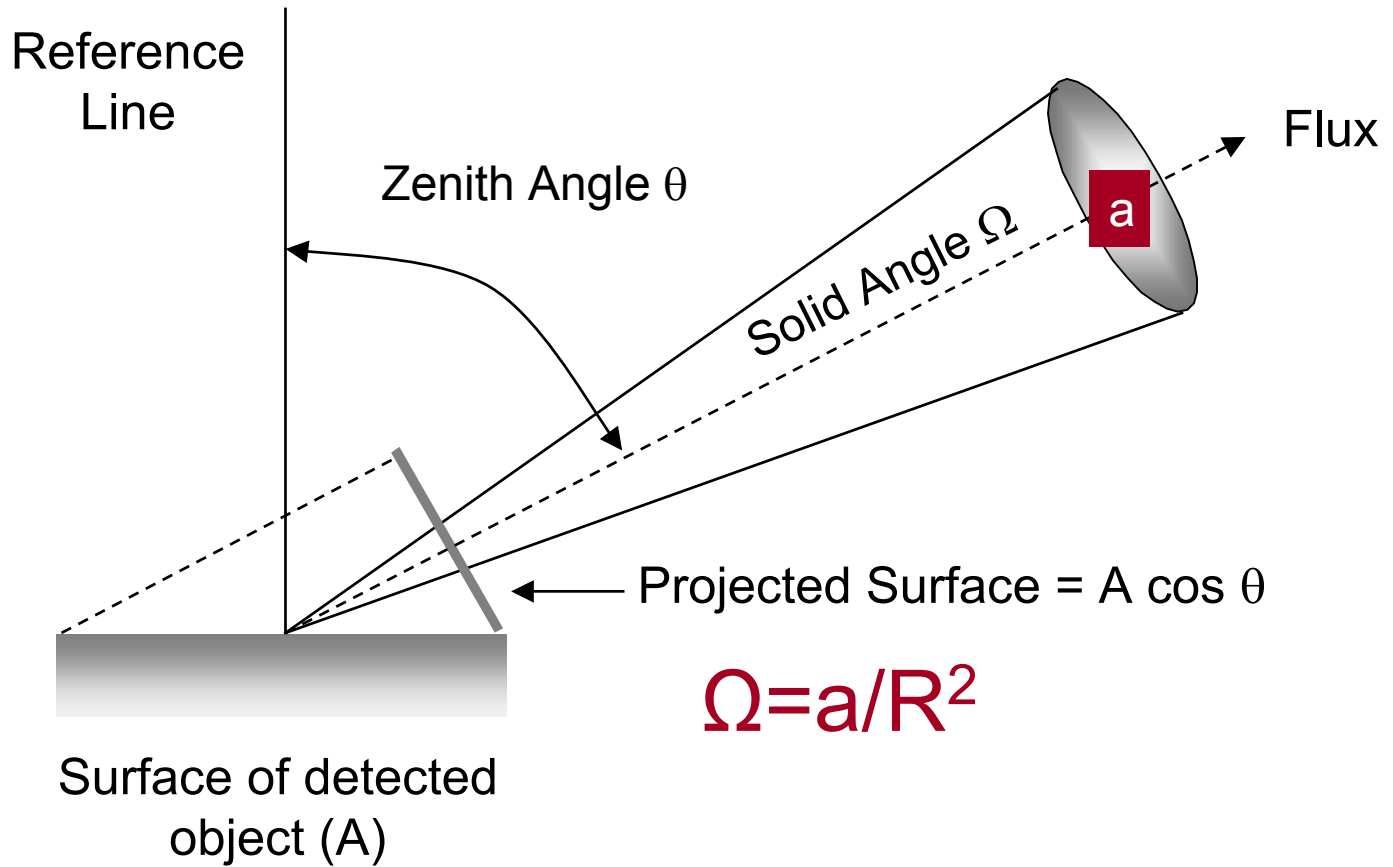
2. Temporal (How Fast Can the System Do?):

Detector and Electronics Responses

3. Resolution of the System (What is the Smallest Temperature the System can resolve?):

NEP

Solid Angle Concept



Radiance L

- Radiance is Defined as the Power per Unit Area per Steradian(Sr)

$$L[\text{W m}^{-2} \text{Sr}^{-1}] = M(T)/\pi$$

Solar Constant K_{solar} (Example)

- Solar disk “subtends” $1/2^\circ$ (or 9 mRadian) in view, the solar constant is the total Radiance Power per unit area
- Since the Radiance is
 $L = 1/\pi M(6000\text{K}) = (\sigma/\pi) \times 6000^4 = 2.34 \times 10^7 \text{ W} \cdot \text{m}^{-2} \text{ Sr}^{-1}$
- The solid angle of the sun is
- $\Omega = (\pi/4)(0.009/2)^2 \sim 6.4 \times 10^{-5} \text{ Sr}$
- The Solar Constant is then:
- $K_{\text{solar}} = L \cdot \Omega \sim 1.5 \text{ KW/M}^2$
- σ : Stefan-Boltzmann’s constant $5.67 \times 10^{-8} \text{ W} \cdot \text{m}^{-2} \cdot \text{K}^{-4}$

Equilibrium Temperature Concept

- The Total Power Absorbed by a 1M^2 Plate Perpendicular to Sun Rays is a Solar Constant K_{solar} of 1.5KW
- The Radiated Power is
- The **Equilibrium Thermodynamic** Condition Stipulates:
- $\sigma T_{\text{plate}}^4 = K_{\text{solar}}$
- $T_{\text{plate}} = (1500/\sigma)^{1/4} \sim 403\text{K} = 130^\circ\text{C}$

$$T_{equi} = \sqrt[4]{\frac{\alpha K_{solar}}{\epsilon \sigma}}$$

How to Manipulate the Equilibrium Temperature T_{equi}

- By varying Surfaces Solar Absorption Coefficient α and ϵ
- For α of 0.2 and ϵ of 0.9, $T_{equi} \sim 277K \Rightarrow 4^\circ C!$
- $\alpha K_{solar} = \epsilon \sigma T_{plate}^4$

$$T_{equi} = \sqrt[4]{\frac{\alpha K_{solar}}{\epsilon \sigma}}$$

- For α of 0.2 and ϵ of 0.9, $T_{equi} \sim 277K \Rightarrow 4^\circ C!$

Why is a Metal Surface so Warm in the Sun?

Polished Metal Surfaces have low α and ϵ

Assume $\alpha = \epsilon = 0.2$

$$T_{equi} = \sqrt[4]{\frac{\alpha K_{solar}}{\epsilon \sigma}} = \sqrt[4]{\frac{0.2 \times 1500}{0.2 \times 5.67 \times 10^{-8}}} = 403K!$$

- σ : Stefan-Boltzmann's constant $5.67 \times 10^{-8} \text{ W} \cdot \text{m}^{-2} \cdot \text{K}^{-4}$

Does “Absolute Temperature” Have to Do with Heat Transfer?

- Conduction

$$\Delta Q \sim \Delta T$$

- Convection

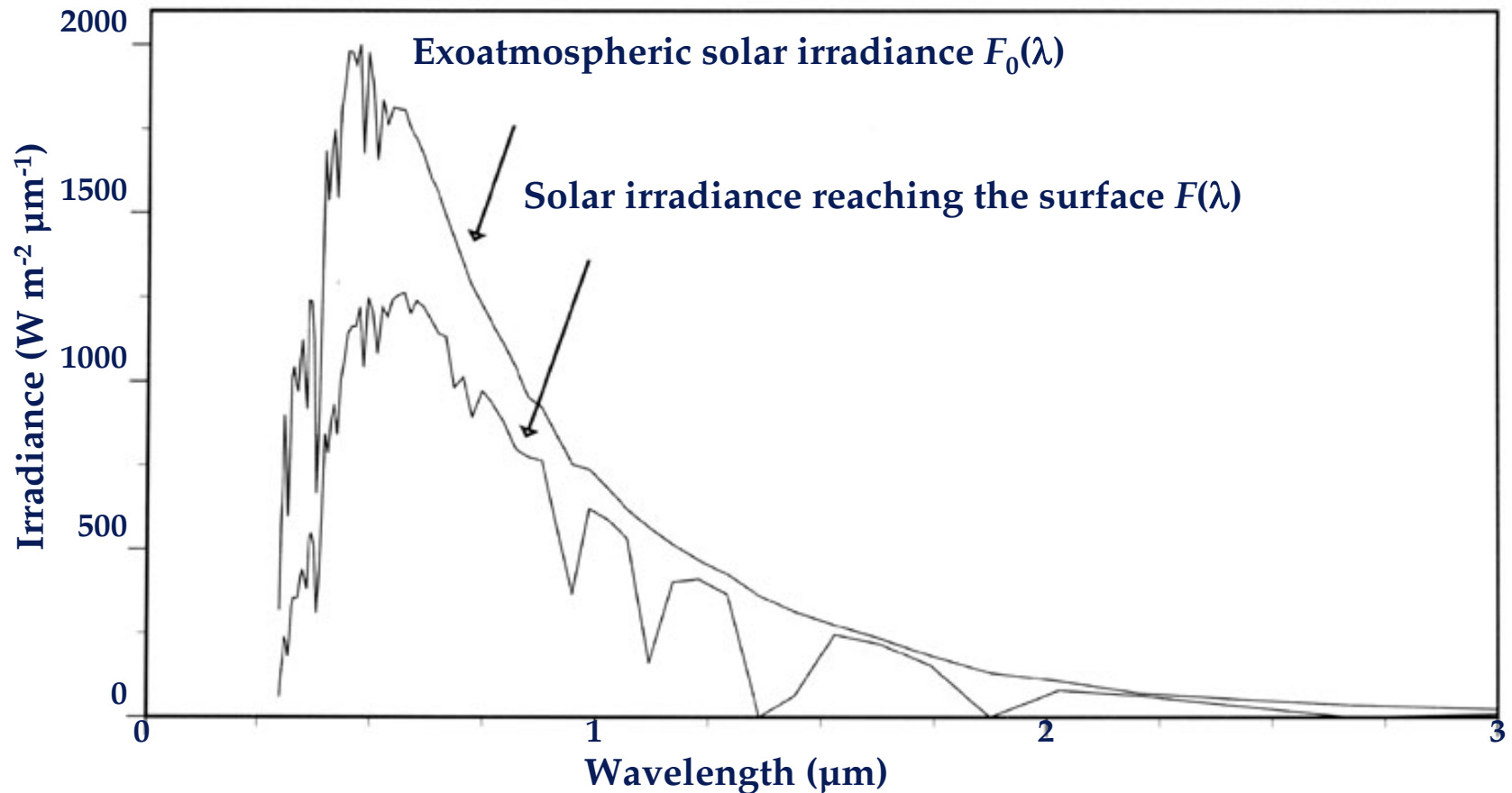
$$\Delta Q \sim \Delta T^n; n \neq 1$$

- Radiation

$$\Delta Q \sim \Delta (T_1^4 - T_2^4)$$

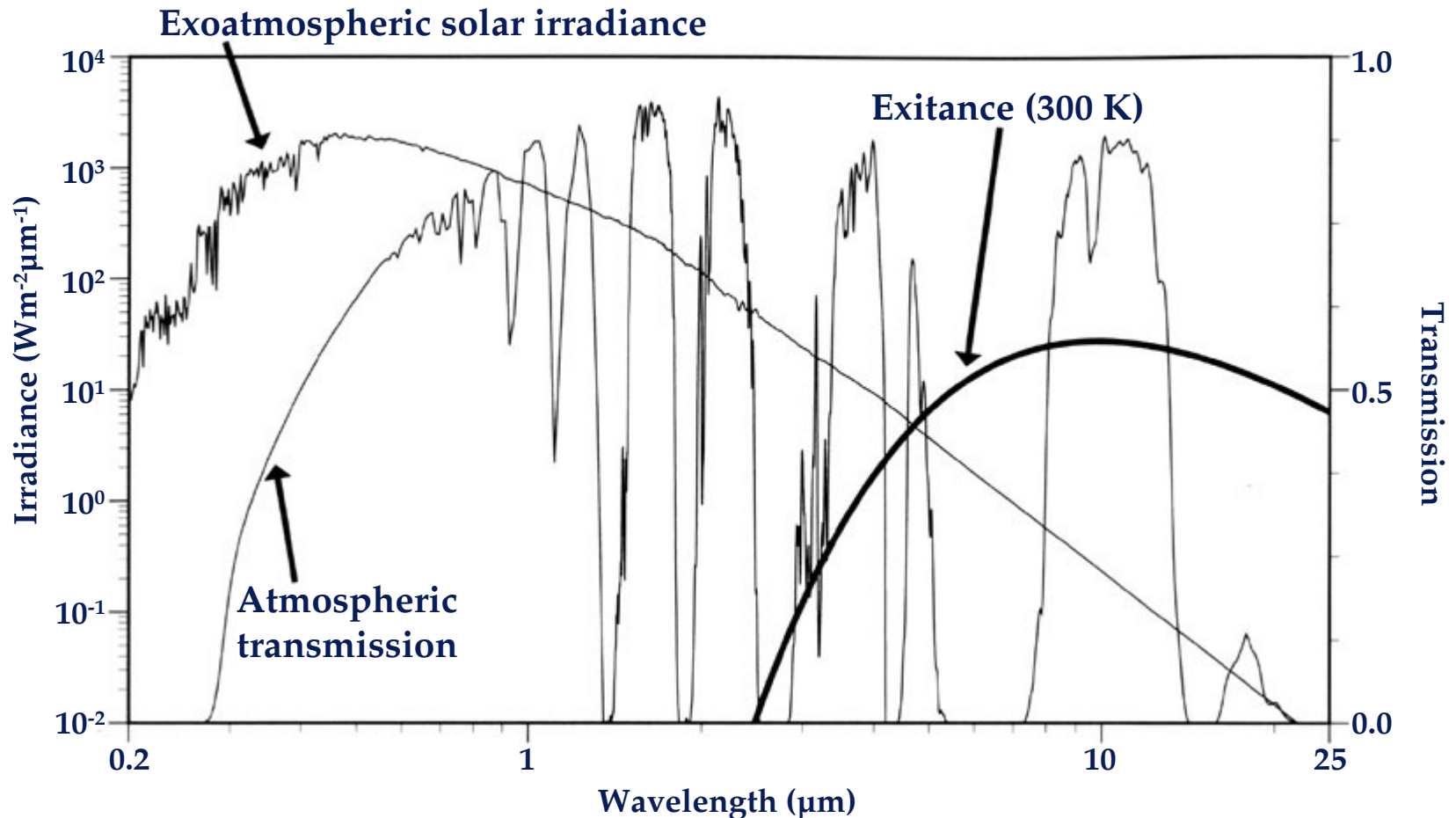
Radiative Heat Transfer is the Only Form of Heat Transfer that requires Absolute Temperature instead of Temperature Difference

Scattering of Sunlight by the Earth- Atmosphere-Surface System



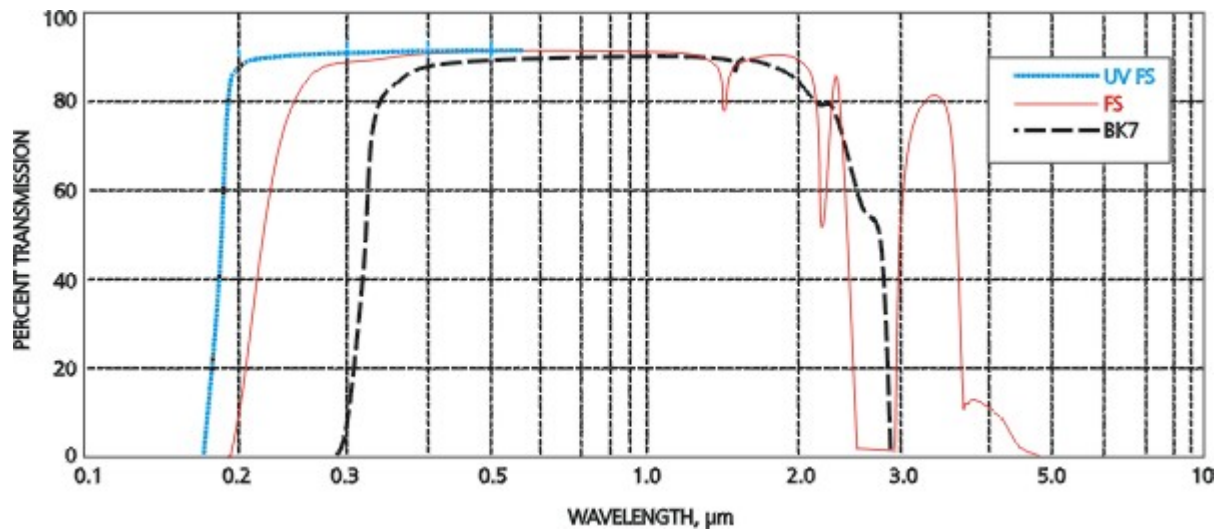
Atmospheric Transmission and Greenhouse Effects

http://tbrs.arizona.edu/education/553-2004/2004/Lect083104_Ch2.ppt-link.ppt#21



BK-7 Transmission Curve

- Most Plate Glass, Similar to BK7
- Plate Glass is Opaque to LWIR

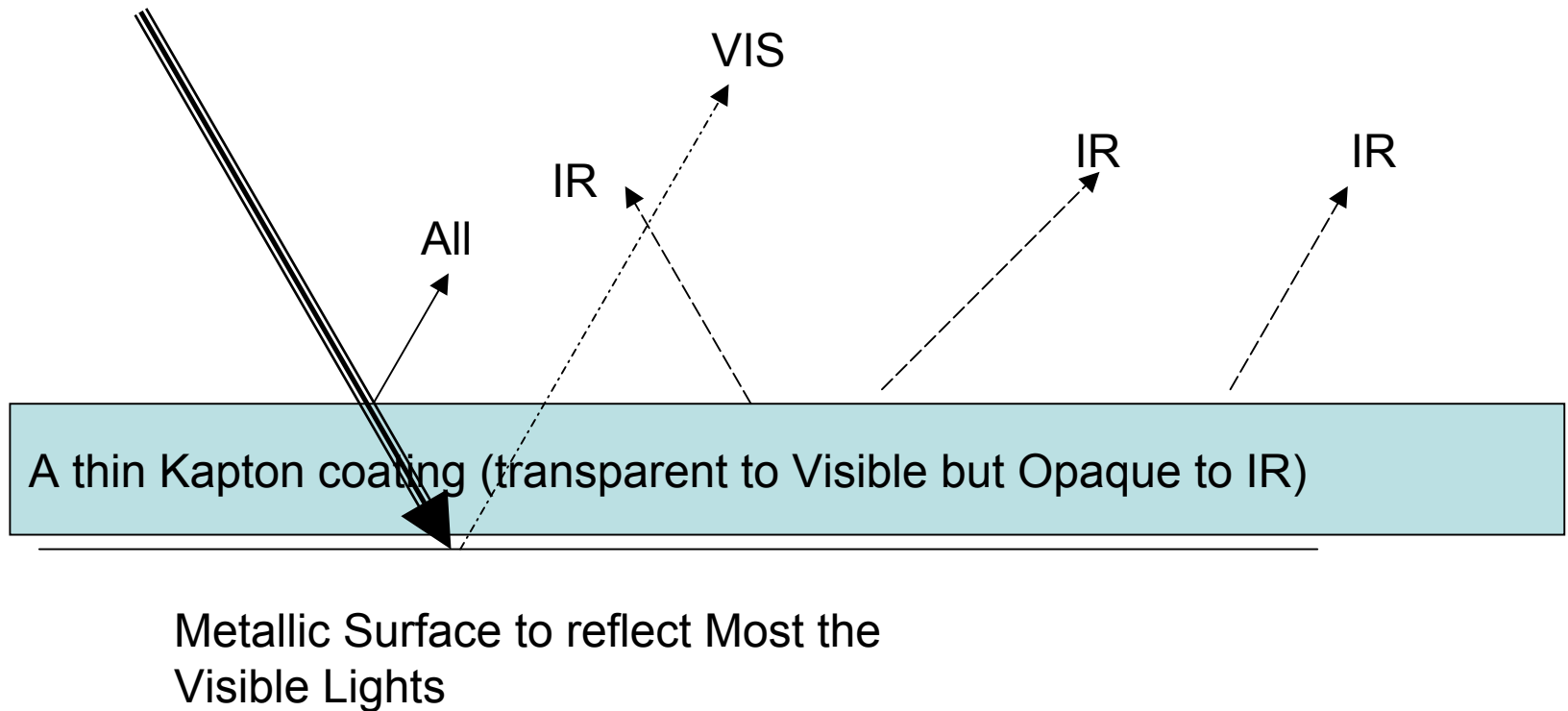


Why is the Interior of a Car so Warm in the Sun?

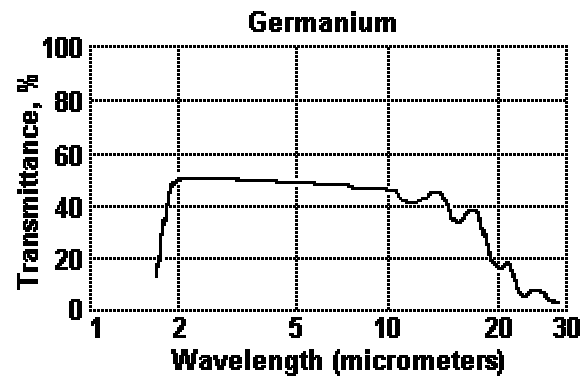
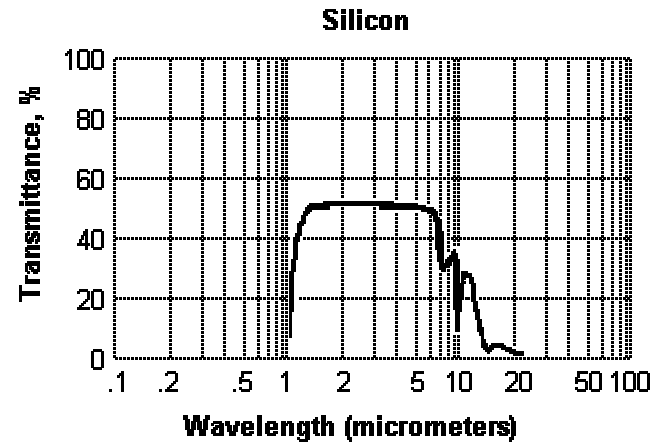
- Sun(6000K) warms a car with all wavelengths, but the interior of the car (300K-400K) emits IR that can not pass through the glass.

So How Does a Space Suit Work in the Sun?

- By a “Secondary Mirror” Surface!

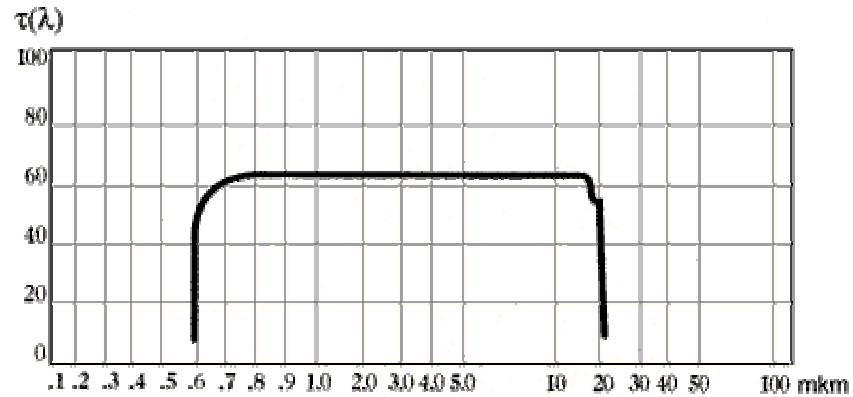


Si and Ge IR Transmissions



ZnSe Transmission

- <http://www.almazoptics.com/ZnSe.html>

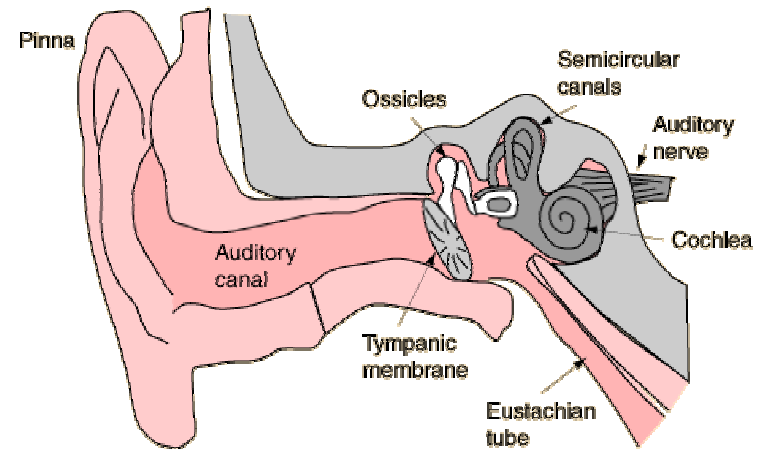


Regardless the “skin tone” difference,
all men are equal in Infrared

- Yes, about 0.98; **almost black!**

What is an Aural Thermometer", or Infrared Aural sensor

- Tympanic cavity as a blackbody cavity
- Emissivity~1.00
- Readily calibrated
- ****Must be in a cavity!!**



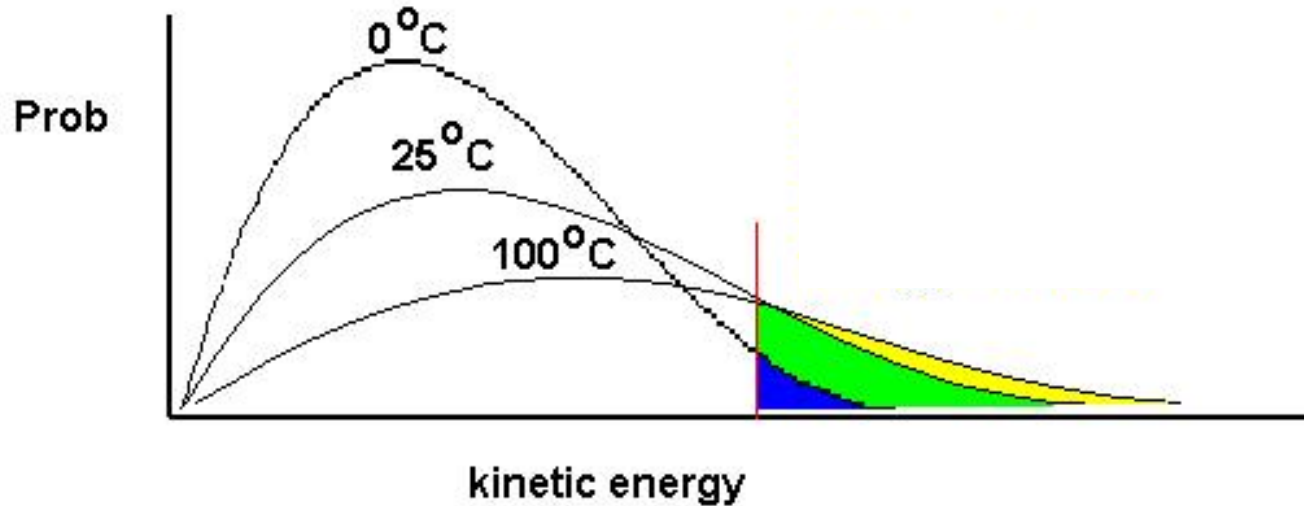
The Infamous SAR Fighter: Ear Cavity Thermometer

- a clinically reliable indicator of **body core** temperature
- Pyro-Electric Transducer



Electron Thermal Energy: Why IR Detectors Must be Cooled!

- $$KE_{avg} = \left[\frac{1}{2} m v^2 \right] = \frac{3}{2} kT$$



NEP Concept

- If we use the entire spectrum, then to detect 38°C (vs. 37 °C), the difference is $[(38 + 273)/(37 + 273)]^4 = 1.013\%$

So to resolve 1°C the “system” must be able to resolve 1.3% difference

=>Noise Equivalent Power or NEP

How good is my System Stacking Against the Others?

D*

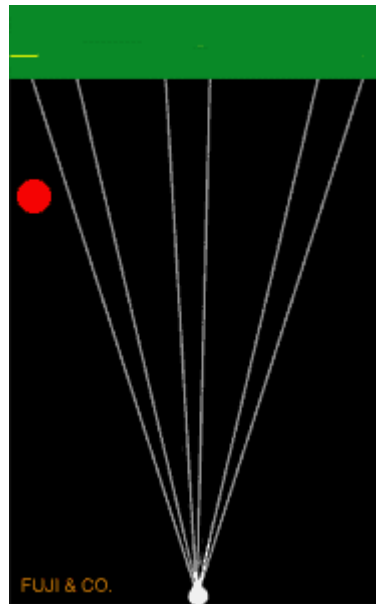
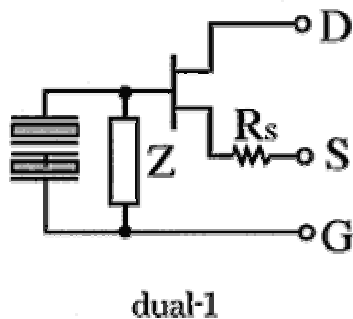
Pyro-electric Detectors

- Pyro: Gk “Fire”
- Pyro-electric: electrical output caused by heat
- Sometimes used for “fiery sparks” display for stage effects
- Low sensitivity, low cost
- Usually for intrusion detection only

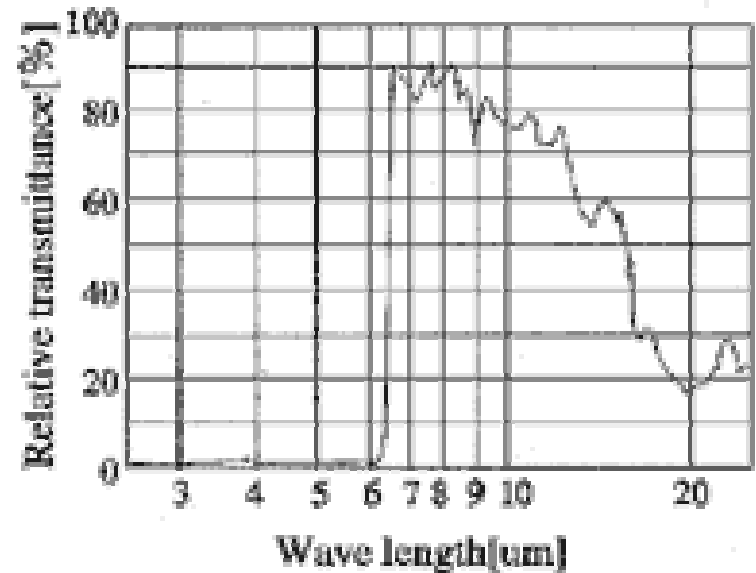
Pyro-electric Detector

polyethylene Fresnel lens are typically used for their low costs

TGS (Tri-glicine-sulfate)



typeB(7 μ m cut-on)



<http://www.fuji-piezo.com/TechGen.htm>

PV $\text{Hg}_x\text{C}_{1-x}\text{Te}$

- Short for “photo-voltaic Mer-Cad-Telluride”, or, “Mer-Cad”
- Chemical compound of HgTe and CdTe
- Response ranging from $1\mu\text{m}$ to $5.5\mu\text{m}$, and $8\mu\text{m}$ up to $13\mu\text{m}$, depending on the Hg to Cd ratio
- Most versatile IR detector

PC HgCTe

- Response to 18 microns
- Need “chopping”
- Response varying with temperature
- Operative in higher temperature than PV

Thermal Transducer is “Export Control” Items

- InSb, HgCdTe, and room-temperature Thermal-pile Focal Plane Arrays (FPA) are all “Strategically sensitive” items